INFORMATION DISCLOSURE CITATION (Use several sheets if negrous)				Docket Number (optional) Application Number 10/705,735					
				Applicant(s) Hampden-Smith et al.					
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- Or	1.	3,676,358	07/11/72	Dale et al.	252	301.4			
	2.	3;691,088	09/12/72	Pelton	252	301.6	<u> </u>		
	3.	3,709,826	01/09/73	Pitt et al.	252	301.4	<u> </u>		
	4.	3,731,353	05/08/73	Vecht	29	25.13			
<u> </u>	5.	-3,742,277	06/26/73	Peters	313	92			
	6.	3,875,449	04/01/75	Byler et al.	313	466			
	7.	3,981,819	09/21/76	Yocom et al.	252	301.4	<u> </u>		
	8.	3,984,586	10/05/76	Kawarada et al.	427	64			
	9.	4,208,299	06/17/80	Oikawa et al.	252	301.6			
	10.	4,208,461	06/17/80	Vanderpool	428	207			
<u> </u>	11.	4,209,567	06/24/80	Takahara et al.	428	403			
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¥	50.	SOLID-STATE SCIEN	NCE AND TECHNO	es of ZnS and (Zn, Cd)S for Pho IOLOGY; June 1988; pp. 1578-	-1583.	•			
G	51.	Faria, S.; Electrolumin 1988; Vol. 135, No. 10	escent Characteristi J; pp. 2627-2630.	ics of Small Particle Size Phosp	hors; Silicon	Epitaxial Film	ns; Octob	er	
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				U.S. PA	ATENT DOCUMENTS					
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6		12.	4,275,333	06/1981	Kagami et al.	252	301.4 S			
		13.	4,287,229	09/01/81	Watanabe et al.	427	64			
		14.	4,339,501	07/13/82	Inoue et al.	428	404			
		15.	4,365,184	12/21/82	Higton et al.	313	503			
		16.	4,374,037	02/15/83	Takahashi	252	301.4			
		17.	4,377,769	03/1983	Beatty et al.	252	301.6 S			
		18.	4,436,646	03/13/84	Takahara et al.	252	301.4			
		19.	4,442,170	04/10/84	Kaule et al.	428	403			
	Û	20.	4,508,760	04/02/85	Olson et al.	427	213.34			
		21.	4,515,827	05/07/85	Dodds et al.	427	68			
		22.	4,724,161	02/09/88	Coutts et al.	427	57			
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Ü	,	52.	136,No. 4; April 1989;	; pp. 1229-1232.	scence Decay Time of CaS:Mn					
(3	53.		rolled Copper-Coat	ting Method For the Preparation 26, pp.399-406; 1991.	of ZnS:Mn	DC Electrolun	ninescent	1	
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_Ω.		23.	4,806,389	02/21/89	Peters et al.	427	67			
-0		24.	4,826,727	05/02/89	Glaser	428	403			
		25.	4,853,254	08/01/89	Wolfe	427	64			
		26.	4,855,189	08/08/89	Simopoulos et al.	428	690			
		27.	4,874,985	10/17/89	Hase et al.	313	487			
		28.	4,902,567	02/20/90	Eilertsen et al.	428	328			
		29.	4,921,727	05/01/90	Datta et al.	427	57			
		30.	4,948,527	08/14/90	Ritsko et al.	252	301.4			
		31.	5,080,928	01/14/92	Klinedinst et al.	427	70			
		32.	5,156,885	10/20/92	Budd	427	70			
		33.	5,244,750	09/14/93	Reilly et al.	428	690			
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<u>_</u>	,	55.			ence of Alkaline Earth Sulphide			6. 198	₆	
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9	34.	5,363,012	11/08/94	Mizukami et al.	313	468		
	35.	5,413,736	05/09/95	Nishisu et al.	252	301.4		
	36.	5,418,062	05/23/95	Budd	428	403		
	37.	5,455,489	10/03/95	Bhargava	315	169.4		
	38.	5,644,193	07/01/97	Matsuda et al.	313	486		
	39.	5,662,831	09/02/97	Chadha	252	301.4		
	40.	5,772,916	06/1998	Jamil et al.	252	301.4R	<u></u>	-
	41.	5,811,924	09/1998	Okumura et al.	313	487		•
	42.	6,039,894	03/2000	Sanjurjo et al.	252	301.48		
	43.	6,132,642	10/2000	Kane	252	301.48		
	44.	6,153,123	11/2000	Hampden-Smith et al.	252	301.48		
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0	56.	Tamaki et al.; Formation	on of CdS Fine Part	ticles by Spray-Pyrolsis; 1995; p	p. 1388-139	0.		
Q	57.	Tohge et al.; Formation Appl. Phys.; Vol. 34 (1		of Zinc Sulfide from Thiourea Copp. 207-209.	omplexes by	Spray Pyroly	sis; Jpn.	J.
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U	45.	6,168,731	01/2001	Hampden-Smith et al.					
	46.	6,193,908	02/2001	Hampden-Smith et al.	252	301.48	<u> </u>		
	47.	6,210,604	04/2001	Hampden-Smith et al.	252	301.48	<u> </u>		
C	48.	6,627,115	09/30/03	Hampden-Smith et al.	252	301.6R	<u> </u>		
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